

## Jigsaw 2E

1. [Hore Section 2.2] How many distinct chemical shifts would you expect to find in the  $^1\text{H}$  spectra of the three isomers of dibromobenzene? *See also: Jigsaws 2A.1, 2B.3 2C.2, and 2D.1.*
2. [Hore Section 2.3] For each of the following sets, rank the options from highest to lowest  $^1\text{H}$  chemical shift. Assume the diamagnetic contribution is the dominant contribution. For each answer, explain why.
  - a.  $\text{CH}_4$ ,  $\text{CH}_3\text{Cl}$ ,  $\text{CH}_2\text{Cl}_2$ , and  $\text{CHCl}_3$ .
  - b. The ortho, meta, and para protons in styrene.
  - c. The protons in  $\text{CH}_2\text{BrCH}_2\text{CH}_3$ .
3. \* [Hore Section 2.1, Week 2 Slide 25, and Keeler Section 4.2] The  $B_0$  field induces precession of magnetization around the z-axis at the Larmor frequency,  $\omega_0 = -\gamma B_0$ . Here we neglect the effect of relaxation.
  - a. Starting with the magnetization along the z-axis, what is the direction of the magnetization of a  $^{13}\text{C}$  site after one second in a 9.4 T magnetic field? The gyromagnetic ratio for  $^{13}\text{C}$  is  $1.071 \times 10^7 \text{ rad}\cdot\text{s}^{-1}\text{T}^{-1}$ .
  - b. Starting with the magnetization along the x-axis, what is the direction of the magnetization of a  $^{13}\text{C}$  site after one second in a 9.4 T magnetic field?